

# Alkyne-Based, Highly Stereo- and Regioselective Synthesis of Stereodefined Functionalized Vinyl Tellurides

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(*Z*)- $\beta$ -Aryltellurovinylphosphonates and (*Z*)- $\beta$ -aryltellurovinyl sulfones were synthesized via the highly stereoselective *anti*-hydrotelluration of 1-alkynylphosphonates and 1-alkynyl sulfones. The configurations of these compounds were characterized via  $^1\text{H}$  NMR spectra or NOESY experiments and by X-ray diffraction analysis; (*E*)- $\beta$ -aryltellurovinyl sulfones were obtained with the reaction of sodium aryltellurate with (*E*)-2-iodovinyl sulfones to confirm the stereochemistry of the above *anti*-hydrotelluration. When the tandem reaction of alkynes with diaryl ditellurides and sodium arylsulfonates was carried out in AcOH/H<sub>2</sub>O (4/1), the corresponding (*E*)- $\beta$ -aryltellurovinyl sulfones were obtained in *one step* in good yields. This reaction is highly regio- and stereoselective and proceeds by using arylsulfonate as the sulfonyl radical precursor and diaryl ditellurides as free radical acceptors. (*E*)-1-Iodo-2-aryltelluroalkenes can be obtained by the *anti*-addition of ArTeI with terminal alkynes in THF. The stereochemistry of compound **17b** was also determined by X-ray diffraction analysis.

## Introduction

Stereoselective synthesis of substituted alkenes has drawn a lot of attention in organic synthesis. Stereo- and regioselective addition of organic reagents to alkynes is one of the most efficient routes to stereodefined alkenes (Figure 1).

Vinyl tellurides are important synthetic intermediates because of their ready transformation to other organic compounds with retention of configuration,<sup>1</sup> e.g., transmetalations to form their corresponding vinyl lithium,<sup>2</sup> copper,<sup>3</sup> zinc,<sup>4</sup> magnesium,<sup>5</sup> calcium,<sup>5</sup> and sodium reagents<sup>5</sup> and their subsequent cross-coupling reaction to form stereodefined substituted alkenes.<sup>6</sup> Recently the synthesis of metal- or heteroatom-substituted vinyl tellurides has received intensive attention because of their

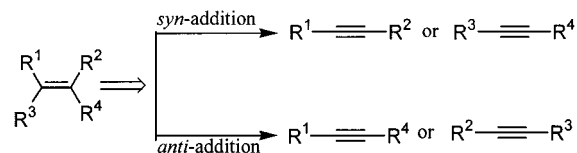


Figure 1.

application in carbon–carbon or carbon-heteroatom bond formation reactions affording polyfunctionalized olefines. Although the preparation of  $\alpha$ -heteroatom-substituted vinyl tellurides **1** and **2** have been extensively studied,<sup>1b,e,7</sup> there are only a few reports on the synthesis of **3**- and

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(1) For reviews about vinylic tellurides, see: (a) Comasseto, J. V.; Ling, L. W.; Petraghani, N.; Stefani, H. A. *Synthesis* **1997**, 373. (b) Dabdoub, M. J.; Begnini, M. L.; Guerrero Jr., P. G. *Tetrahedron* **1998**, *54*, 2371. (c) Dabdoub, M. J.; Justino, A.; Guerrero Jr., P. G. *Organometallics* **1998**, *17*, 1901. (d) Dabdoub, M. J.; Baroni C. M. J. *Org. Chem.* **2000**, *65*, 54–60. (e) Dabdoub, M. J.; Begnini, M. L.; Guerrero, P. G., Jr.; Baroni, C. M. J. *Org. Chem.* **2000**, *65*, 61–67.

(2) (a) Kauffman, T.; Allers, H. *Chem. Ber.* **1983**, *116*, 1001. (b) Hiuro, T.; Mogami, T.; Kambe, N.; Fujiwara, S. I.; Sonoda, N. *Angew. Chem., Int. Ed. Engl.* **1987**, *26*, 1187. (c) Barros, S. M.; Dabdoub, M. J.; Dabdoub, V. M. B.; Comasseto, J. V. *Organometallics* **1989**, *8*, 1661. (d) Barros, S. M.; Comasseto, J. V.; Berriel, J. *Tetrahedron Lett.* **1989**, *30*, 7353. (e) Dabdoub, M. J.; Dabdoub, V. B.; Comasseto, J. V. *Tetrahedron Lett.* **1992**, *33*, 2261. (f) Ogawa, A.; Tsuboi, Y.; Obayashi, R.; Yokoyama, K.; Ryu, I.; Sonoda, N. *J. Org. Chem.* **1994**, *59*, 1600. (g) Dabdoub, M. J.; Dabdoub, V. B. *Tetrahedron* **1995**, *51*, 9839. (h) Dabdoub, M. J.; Begnini, M. L.; Cassol, T. M.; Guerrero Jr., P. G.; Silveira, C. C. *Tetrahedron Lett.* **1995**, *36*, 7623. (i) Mo, X.-S.; Huang, Y.-Z. *Tetrahedron Lett.* **1995**, *36*, 3539. (j) Dabdoub, M. J.; Dabdoub, V. B.; Pereira, M. A.; Zukerman-Schpector, J. *J. Org. Chem.* **1996**, *61*, 9503. (k) Dabdoub, M. J.; Dabdoub, V. B.; Guerrero Jr., P. G.; Silveria, C. C. *Tetrahedron* **1997**, *53*, 4199. (l) Huang, Y.-Z.; Mo, X.-S. *Tetrahedron Lett.* **1998**, *39*, 1945. (m) Dabdoub, M. J.; Jacob, R. G.; Ferreira, J. T. B.; Dabdoub, V. B.; Marques, F. *Tetrahedron Lett.* **1999**, *40*, 7159.

(3) (a) Comasseto, J. V.; Berriel, J. N. *Synth. Commun.* **1990**, *20*, 1681. (b) Tucci, F. C.; Chieffi, A.; Comasseto, J. V. *Tetrahedron Lett.* **1992**, *33*, 5721. (c) Marino, J. P.; Tucci, F. C.; Comasseto, J. V. *Synlett* **1993**, 761. (d) Chieffi, A.; Comasseto, J. V. *Tetrahedron Lett.* **1994**, *35*, 4063. (e) Araujo, M.; Comasseto, J. V. *Synlett* **1995**, 1145. (f) Tucci, F. C.; Chieffi, A.; Comasseto, J. V.; Marino, J. P. *J. Org. Chem.* **1996**, *61*, 4975. (g) Araujo, M. A.; Barrientos-Astigarrá, R. M.; Comasseto, J. V. *Tetrahedron Lett.* **1999**, *40*, 5115.

(4) Terao, J.; Kambe, N.; Sonoda, N. *Tetrahedron Lett.* **1996**, *37*, 4741.

(5) Kanda, T.; Sugino, T.; Kambe, N.; Sonoda, N. *Phosphorus, Sulfur, Silicon Relat. Elem.* **1992**, *67*, 103.

(6) (a) Uemura, S.; Fukuzawa, S. I.; Patil, S. R. *J. Organomet. Chem.* **1983**, *243*, 9. (b) Chieffi, A.; Comasseto, J. V. *Tetrahedron Lett.* **1994**, *35*, 4063. (c) Chieffi, A.; Comasseto, J. V. *Synlett* **1995**, 671. (d) Zeni, G.; Comasseto, J. V. *Tetrahedron Lett.* **1999**, *40*, 4619. (e) Dabdoub, M. J.; Dabdoub, V. B.; Marino, J. P. *Tetrahedron Lett.* **2000**, *41*, 433. (f) Dabdoub, M. J.; Dabdoub, V. B.; Marino, J. P. *Tetrahedron Lett.* **2000**, *41*, 437.

(7) For the preparation of  $\alpha$ -hetero (S, Se, Te, Zr, I, Br, Sn, B)-substituted vinyl tellurides and their synthetic application, see: (a) Ogawa, A.; Ogawa, I.; Obayashi, R.; Umezū, K.; Doi, M.; Hirao, T. *J. Org. Chem.* **1999**, *64*, 86. (b) Braga, A. L.; Zeni, G.; Andrade, L. H.; Silveira, C. C. *Synlett* **1997**, 595. (c) Braga, A.; Reckziegel, A.; Silveira, C. C.; Comasseto, J. V. *Synth. Commun.* **1994**, *24*, 1165. (d) Dabdoub, M. J.; Cassol, T. M.; Barbosa, S. L. *Tetrahedron Lett.* **1996**, *37*, 831. (e) Stang, P. J.; Roberts, K. A.; Lynch, L. E. *J. Org. Chem.* **1984**, *49*, 1653. (f) Silveira, C. C.; Perin, G.; Braga, A. L. *Tetrahedron Lett.* **1995**, *36*, 7361. (g) Park, C. P.; Sung, J. W.; Oh, D. Y. *Synlett* **1999**, 1055. (h) Huang, X.; Liang, C.-G. *Synth. Commun.* **2000**, *30*, 1903.

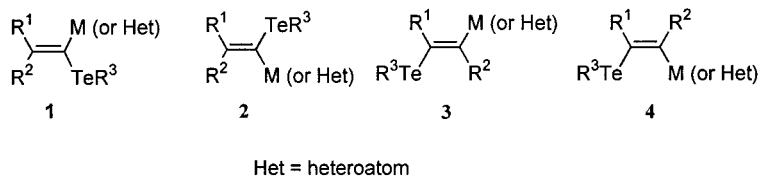
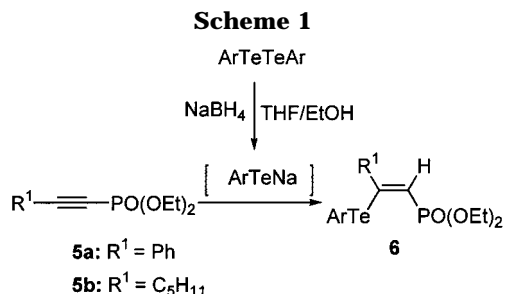


Figure 2.



4-type of highly  $\beta$ -functionalized vinylic tellurides<sup>2f,8,9</sup> (Figure 2).

As a part of our ongoing study aimed at the synthesis and application of bifunctionalized vinyl tellurides,<sup>7h,9a-c</sup> we report herein the stereo- and regioselective synthesis of vinyltellurides containing sulfonyl, phosphonyl, and iodo groups via the stereo- and regiodefined addition of organic tellurides with alkynes.

## Results and Discussion

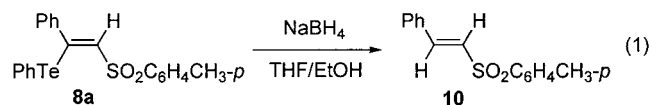
**Highly Stereo- and Regioselective *Anti*-Hydro-telluration of 1-Alkynylphosphonates and 1-Alkynyl Sulfones.** Hydrotelluration of acetylenes is a widely used method for the synthesis of vinyl tellurides. In the case of addition to acetylenes bearing electron-withdrawing groups such as acetylenic aldehydes, ketones, carboxylic acids, and esters,  $\beta$ -carbonylvinyl tellurides with *Z*-configuration can be obtained.<sup>10</sup> As our work was initiated, there was no report on the hydrotelluration of 1-alkynyl sulfones and 1-alkynylphosphonates. First, we investigated the reaction of 1-alkynylphosphonate **5a** (R<sup>1</sup> = Ph) with PhTeNa (prepared by the reduction of ditelluride with NaBH<sub>4</sub>) in THF/EtOH (1/1) at room temperature (monitored by TLC). This reaction afforded *anti*-hydrotelluration product, i.e., *Z*-2-phenyltellurovinylphosphonate **6c**, stereo- and regioselectively as determined by the <sup>1</sup>H NMR spectrum of the crude product. Other substrates gave the same results (Scheme 1 and Table 1).

The structures of these compounds were affirmatively characterized via <sup>1</sup>H NMR spectra or NOESY experiments (*Z*-**6a** and *Z*-**6e**) and by X-ray diffraction analysis of compound **6a** to be in *Z*-configuration. In all cases, the attack of sodium tellurate on alkynes occurred at the  $\beta$ -position exclusively.

When the reaction of 1-alkynyl sulfones with ArTeNa was carried out at room temperature for 20 min, the *Z*-2-

aryltelluro-1-alkenyl sulfones (**8**) were obtained in good yields (Scheme 2 and Table 1). No isomeric compounds were obtained as determined by the <sup>1</sup>H NMR spectra of the crude products. The *Z*-configurations of these compounds were characterized via the coupling constant of vinyl protons (*J* = 8.69 Hz) of compound **9a** and the NOE experiments for compounds **8b** and **9b**. The correlations between the vinyl hydrogen and aryl proton in **8b** or the vinyl hydrogen and allylic hydrogen in **9b** indicated that the vinyl hydrogen is situated close to those groups (*cis* relation) (Figure 3).

However, when the reaction mixture was stirred for a longer time (overnight), **8a** was easily converted to (*E*)-2-phenylvinyl sulfone **10** in 76% yield via reductive detelluration (eq 1).



To further confirm the stereochemistry of the hydro-telluration of 1-alkynyl sulfones **7**, we developed a way to reach *E*-2-aryltelluryl-1-alkenyl sulfones by the reaction of *E*-2-iodo-1-alkenyl sulfones with sodium aryltellurates in THF/EtOH at 0 °C. In all cases studied, only the *E*-isomers (Scheme 3 and Table 2), which were characterized by IR, MS, and <sup>1</sup>H and <sup>13</sup>C NMR spectroscopy and comparison with the corresponding *Z*-isomers, were obtained.

On the other hand, for compound *E*-**13f**, there was no NOE correlation between the vinyl proton (6.73 ppm) and the allylic hydrogen (3.55–3.58 ppm). NOE correlation between the vinyl proton and aryl proton of the aryltellurium group was observed. This indicated that vinyl proton situated close to aryltellurium group (*trans* relation) (compare with compound *Z*-**9b**) (Figure 3).

***Anti*-Tellurosulfonation of 1-Alkynes.** It was reported that arylselenovinyl sulfones can be prepared conveniently from the selenosulfonation of alkynes with PhSeSO<sub>2</sub>Ar.<sup>11</sup> We have developed here the stereo- and regioselective synthesis of *E*-2-aryltelluryl-1-alkenyl sulfones **12** by the tellurosulfonation reaction of alkynes with ArTeTeAr and sodium arylsulfonates in AcOH/H<sub>2</sub>O at 80 °C. The results of the reaction of alkynes (1.1 mmol) with diaryl ditellurides (0.5 mmol) and ArSO<sub>2</sub>Na (5 mmol) in AcOH/H<sub>2</sub>O at 80 °C are shown in Table 3. With R = aryl, the reaction finished within 5 h, while with R = alkyl, the reaction was slower and finished within 24 h (Scheme 4). The reaction of trimethylsilylacetylene with PhTeTePh and sodium arylsulfonate in AcOH/H<sub>2</sub>O for 48

(8) (a) Moura Campos, M.; Petragnani, N. *Tetrahedron* **1962**, *18*, 527. (b) Uemura, S.; Miyoshi, H.; Okano, M. *Chem. Lett.* **1979**, 1357. (d) Stefani, H. A.; Petragnani, N.; Zukerman-Schpector, J.; Dornelles, L.; Silva, D. O.; Braga, A. L. *J. Organomet. Chem.* **1998**, *562*, 127.

(9) (a) Huang, X.; Wang, Y.-P. *Tetrahedron Lett.* **1996**, *37*, 7417. (b) Huang, X.; Wang, Y.-P. *Synth. Commun.* **1996**, *26*, 3087. (c) Zhao, C.-Q.; Huang, X. *Synth. Commun.* **1997**, *27*, 237.

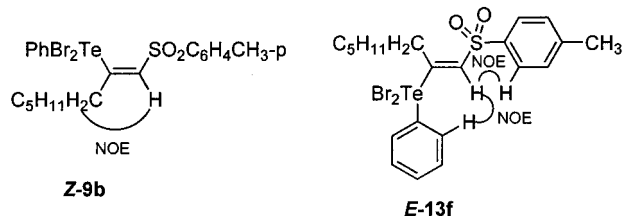
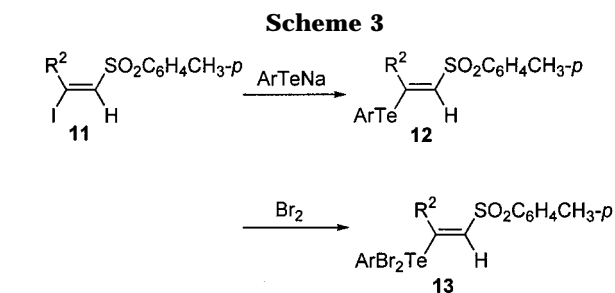
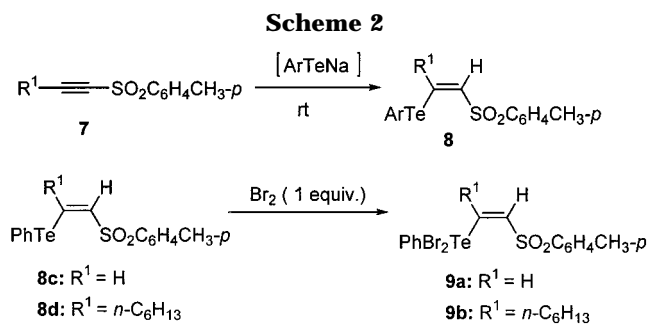
(10) (a) Detty, M. R.; Murray, B. J.; Smith, D. L.; Zumbulyadis, N. *J. Am. Chem. Soc.* **1983**, *105*, 875. (b) Detty, M. R.; Murray, B. J. *J. Am. Chem. Soc.* **1983**, *105*, 883.

(11) (a) Back, T. G.; Collins, S.; Kerr, R. S. *J. Org. Chem.* **1983**, *48*, 3077. (b) Back, T. G.; Collins, S.; Gokhale, U.; Law, K.-W. *J. Org. Chem.* **1983**, *48*, 4776. (c) Miura, T.; Kobayashi, M. *J. Chem. Soc., Chem. Commun.* **1982**, 438. (d) Back, T. G.; Krishna, M. V. *J. Org. Chem.* **1987**, *52*, 4265. (e) Back, T. G.; Krishna, M. V.; Muralidharan, K. R. *J. Org. Chem.* **1989**, *54*, 4146. (f) Gancarz, R. A.; Kice, J. L. *J. Org. Chem.* **1981**, *46*, 4899.

**Table 1. Synthesis of *Z*-2-aryltelluro-1-alkenylphosphonates 6a–f, *Z*-2-aryltelluro-1-alkenyl sulfones 8a,b, or Dibromides 9a,b**

| entries | R <sup>1</sup>                 | E  | Ar   | reaction condition | product: yield (%)            |
|---------|--------------------------------|--|--|--------------------|-------------------------------|
| 1       | C <sub>6</sub> H <sub>5</sub>  | PO(OEt) <sub>2</sub>   | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | rt, overnight      | <i>Z</i> -6a: 59 <sup>a</sup> |
| 2       | C <sub>6</sub> H <sub>5</sub>  | PO(OEt) <sub>2</sub>   | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | rt, overnight      | <i>Z</i> -6b: 63 <sup>a</sup> |
| 3       | C <sub>6</sub> H <sub>5</sub>  | PO(OEt) <sub>2</sub>   | C <sub>6</sub> H <sub>5</sub>                    | rt, overnight      | <i>Z</i> -6c: 61 <sup>a</sup> |
| 4       | C <sub>6</sub> H <sub>5</sub>  | PO(OEt) <sub>2</sub>   | 4-F-C <sub>6</sub> H <sub>4</sub>                | rt, overnight      | <i>Z</i> -6d: 67 <sup>a</sup> |
| 5       | C <sub>5</sub> H <sub>11</sub> | PO(OEt) <sub>2</sub>   | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | rt, overnight      | <i>Z</i> -6e: 57 <sup>a</sup> |
| 6       | C <sub>5</sub> H <sub>11</sub> | PO(OEt) <sub>2</sub>   | 4-F-C <sub>6</sub> H <sub>4</sub>                | rt, overnight      | <i>Z</i> -6f: 59 <sup>a</sup> |
| 7       | Ph                             | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> SO <sub>2</sub> | Ph   | rt, 20 min         | <i>Z</i> -8a: 74 <sup>b</sup> |
| 8       | Ph                             | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> SO <sub>2</sub> | 4-F-C <sub>6</sub> H <sub>4</sub>                | rt, 20 min         | <i>Z</i> -8b: 75 <sup>b</sup> |
| 9       | H                              | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> SO <sub>2</sub> | Ph   | rt, 20 min         | <i>Z</i> -9a: 82 <sup>c</sup> |
| 10      | C <sub>6</sub> H <sub>13</sub> | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> SO <sub>2</sub> | Ph   | rt, 20 min         | <i>Z</i> -9b: 71 <sup>c</sup> |

<sup>a</sup> Isolated by preparative TLC eluted with AcOEt/hexanes = 1/5. <sup>b</sup> Isolated by chromatography eluted with AcOEt/hexanes = 1/10. <sup>c</sup> Isolated by recrystallization with HCCl<sub>3</sub>/CH<sub>3</sub>OH.

**Figure 3.**

h produced the desilylated *E*-2-phenyltelluro-1-alkenyl sulfone (**12k**) ( $J = 15.82$  Hz)<sup>12</sup> in 26% yield after a workup with aqueous sodium bicarbonate followed by chromatography on silica gel (see Scheme 5 and entry **26**, Table 3). The *Z*-isomer was not obtained. The configurations of other compounds were determined by <sup>1</sup>H NMR, IR, and MS spectroscopy and comparison with the products which were obtained by the reaction of *E*-2-iodo-1-alkenyl sulfones with sodium aryltellurates. No significant amounts of stereoisomers of the above products could be isolated.

Recent reports have shown sulfonyl radical can be generated from sodium arylsulfinate in aqueous acetic acid.<sup>13</sup> A plausible mechanism for the formation of *E*-2-aryltelluro-1-alkenyl sulfones **12** is the regioselective

(12) Desilylation reaction of a vinylsilane with a similar structure in HOAc was known; see: Ohnuma, T.; Heta, N.; Fujiwara, H.; Ban, Y. *J. Org. Chem.* **1982**, *47*, 4713.

**Table 2. Stereoselective Synthesis of *E*-2-Aryltelluro-1-alkenyl Sulfone 12g and Dibromides 13a–f (method A)**

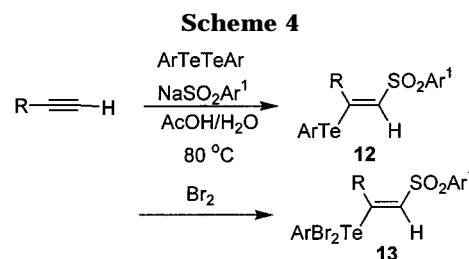
| entry | R                              | Ar                               | product: <sup>a</sup> yield (%) |
|-------|--------------------------------|----------------------------------|---------------------------------|
| 11    | C <sub>4</sub> H <sub>9</sub>  | C <sub>6</sub> H <sub>5</sub>    | <i>E</i> -13a: 72 <sup>c</sup>  |
| 12    | C <sub>4</sub> H <sub>9</sub>  | 4-FC <sub>6</sub> H <sub>4</sub> | <i>E</i> -13b: 71 <sup>c</sup>  |
| 13    | C <sub>5</sub> H <sub>11</sub> | C <sub>6</sub> H <sub>5</sub>    | <i>E</i> -13c: 64 <sup>c</sup>  |
| 14    | C <sub>5</sub> H <sub>11</sub> | 4-FC <sub>6</sub> H <sub>4</sub> | <i>E</i> -13d: 63 <sup>c</sup>  |
| 15    | C <sub>6</sub> H <sub>13</sub> | C <sub>6</sub> H <sub>5</sub>    | <i>E</i> -13e: 57 <sup>c</sup>  |
| 16    | C <sub>6</sub> H <sub>13</sub> | 4-FC <sub>6</sub> H <sub>4</sub> | <i>E</i> -13f: 62 <sup>c</sup>  |
| 17    | C <sub>6</sub> H <sub>5</sub>  | C <sub>6</sub> H <sub>5</sub>    | <i>E</i> -12g: 63 <sup>b</sup>  |

<sup>a</sup> Reaction conditions: 0 °C, 20 min. <sup>b</sup> Isolated yields by chromatography on silica gel eluted with AcOEt/hexanes = 1/10. <sup>c</sup> The yield of dibromotelluranyl compounds by recrystallization from HCCl<sub>3</sub>/CH<sub>3</sub>OH.

**Table 3. Synthesis of *E*-2-aryltelluro-1-alkenyl sulfones 12g, 12h, and 12k or Dibromides 13 from Tellurosulfonation of 1-Alkynes (method B)**

| entry | R                              | Ar                                | Ar'  | reaction time (h) | product: yield <sup>a</sup> (%) |
|-------|--------------------------------|-----------------------------------|--|-------------------|---------------------------------|
| 18    | C <sub>4</sub> H <sub>9</sub>  | C <sub>6</sub> H <sub>5</sub>     | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | 24                | <i>E</i> -13a: 55               |
| 19    | C <sub>5</sub> H <sub>11</sub> | C <sub>6</sub> H <sub>5</sub>     | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | 24                | <i>E</i> -13c: 70               |
| 20    | C <sub>5</sub> H <sub>11</sub> | 4-F-C <sub>6</sub> H <sub>4</sub> | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | 24                | <i>E</i> -13d: 53               |
| 21    | C <sub>6</sub> H <sub>13</sub> | C <sub>6</sub> H <sub>5</sub>     | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | 24                | <i>E</i> -13e: 68               |
| 22    | C <sub>6</sub> H <sub>5</sub>  | C <sub>6</sub> H <sub>5</sub>     | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | 5                 | <i>E</i> -12g: 74               |
| 23    | C <sub>6</sub> H <sub>5</sub>  | C <sub>6</sub> H <sub>5</sub>     | 4-Cl-C <sub>6</sub> H <sub>4</sub>               | 2                 | <i>E</i> -12h: 85               |
| 24    | C <sub>4</sub> H <sub>9</sub>  | C <sub>6</sub> H <sub>5</sub>     | C <sub>6</sub> H <sub>5</sub>                    | 24                | <i>E</i> -13i: 51               |
| 25    | C <sub>6</sub> H <sub>13</sub> | C <sub>6</sub> H <sub>5</sub>     | C <sub>6</sub> H <sub>5</sub>                    | 24                | <i>E</i> -13j: 61               |
| 26    | Me <sub>3</sub> Si             | C <sub>6</sub> H <sub>5</sub>     | 4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> | 48                | <i>E</i> -12k: 26 <sup>b</sup>  |

<sup>a</sup> Isolated yield based on ArTeTeAr. <sup>b</sup> The yield of desilylated product.

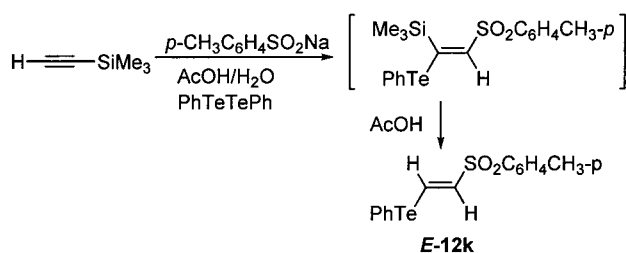


addition of a sulfonyl radical to the alkyne to form the 2-sulfonylvinyl radical intermediates **14** and **15**,<sup>11a</sup> further reaction of **14** with diaryl ditelluride produces products **12** highly stereoselectively probably due to the bigger steric interaction of SO<sub>2</sub>Ar<sup>1</sup> in **15** with the approaching ArTeTeAr (Scheme 6).<sup>14</sup>

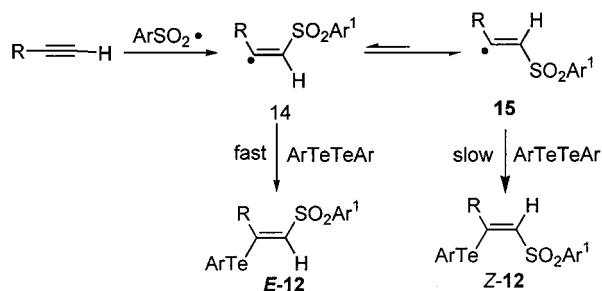
**Anti-Addition of ArTeI with 1-Alkynes.** Recent reports by our group and others have shown that the

(13) Chuang, C. P.; Wang, S. F. *Synth. Commun.* **1995**, *25*, 3549, and references therein.

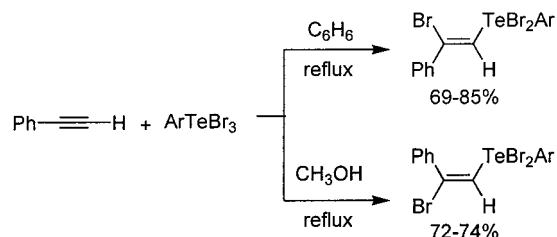
Scheme 5



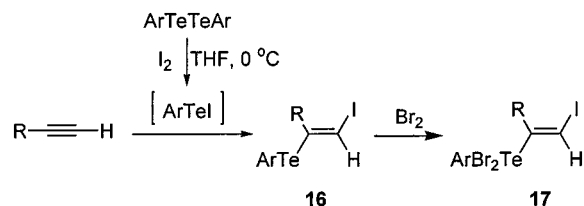
Scheme 6



Scheme 7



Scheme 8



addition reaction of alkynes with  $\text{ArTeX}_3$  ( $\text{X} = \text{Cl}, \text{Br}$ ) occurred stereo- and regioselectively. We have found the reaction of aryltellurium tribromides with alkynes gave *E*- or *Z*-2-bromovinyl aryl tellurium dibromides, depending on the solvents used.<sup>9a</sup> Further studies involve their application to the synthesis of polysubstituted alkenes (Scheme 7).<sup>9a,b</sup>

It was also been reported that aryltelluranyl iodides can react as good electrophilic reagents with alkenes<sup>15</sup> or alkynylboronates.<sup>16</sup> All these promoted us to examine the reaction of aryltelluranyl iodides with alkynes. It was found that aryltelluranyl iodides reacted stereo- and regioselectively with alkynes in THF to afford *E*-1-iodo-2-aryltelluro-1-alkenes **16**, which can react with  $\text{Br}_2$  to give the more stable dibromotellurides **17** with the retention of the configuration (Scheme 8 and Table 4). The structure of product **17b** was clearly established by an X-ray diffraction analysis. Reaction of aryltelluranyl iodides with alkynes in benzene and  $\text{CH}_2\text{Cl}_2$  was not clean.

Table 4. Synthesis of (2-Iodo-1-alkyl)vinyl Aryl Tellurium Dibromides 17a–f

| entry | Ar                          | R                         | product: <sup>a,b</sup> (yield%) |
|-------|-----------------------------|---------------------------|----------------------------------|
| 27    | $\text{C}_6\text{H}_5$      | $\text{C}_4\text{H}_9$    | <i>E</i> -17a: 45                |
| 28    | $4\text{-Cl-C}_6\text{H}_4$ | $\text{C}_4\text{H}_9$    | <i>E</i> -17b: 53                |
| 29    | $\text{C}_6\text{H}_5$      | $\text{C}_5\text{H}_{11}$ | <i>E</i> -17c: 48                |
| 30    | $4\text{-Cl-C}_6\text{H}_4$ | $\text{C}_5\text{H}_{11}$ | <i>E</i> -17d: 51                |
| 31    | $\text{C}_6\text{H}_5$      | $\text{C}_6\text{H}_{13}$ | <i>E</i> -17e: 37                |
| 32    | $4\text{-Cl-C}_6\text{H}_4$ | $\text{C}_6\text{H}_{13}$ | <i>E</i> -17f: 43                |

<sup>a</sup> Reaction conditions: the reaction was carried out using  $\text{ArTeTeAr}$  (0.5 mmol),  $\text{I}_2$  (0.55 mmol), and alkyne (1 mmol) in THF at room temperature for 24 h. <sup>b</sup> Isolated yields by recrystallization from  $\text{HCCl}_3/\text{CH}_3\text{OH}$ .

## Conclusion

*Anti*-hydrotelluration of 1-alkynyl sulfones or 1-alkynylphosphonates afforded *Z*-2-aryltelluro-1-alkenyl sulfone or phosphonates exclusively; *E*-2-aryltelluro-1-alkenyl sulfones were obtained by the reaction of aryltelluro-1-alkenyl sulfones. When the reaction of alkynes with diaryl ditelluride and sodium arylsulfinate was carried in  $\text{AcOH}/\text{H}_2\text{O}$  (4:1), the corresponding *E*-2-aryltelluro-1-alkenyl sulfones were obtained in good yields stereo- and regioselectively.  $\text{ArTeI}$  reacted with alkynes in THF to afford *E*-1-iodo-2-aryltelluroalkenes stereo- and regioselectively. Since vinyl sulfones,<sup>17</sup> vinylphosphonates,<sup>18</sup> and vinyl iodides<sup>19</sup> are important intermediates in organic synthesis, we envisaged that these three classes of compounds are intermediates of synthetic importance with the combination of the well-known chemical reactivity of the carbon–tellurium bonds.

## Experimental Section

**General.** <sup>1</sup>H NMR spectra were recorded on a 400 or 500 MHz spectrometer and all <sup>13</sup>C NMR spectra were recorded on a 400 MHz spectrometer using  $\text{CDCl}_3$  as the solvent. Diaryl ditellurides,<sup>10a</sup> 1-alkynyl sulfones,<sup>20</sup> 1-alkynylphosphonates,<sup>21</sup> and *E*-2-iodovinylsulfonates<sup>22</sup> were prepared by the methods reported in the literature.

***Anti*-Hydrotelluration of 1-Alkynylphosphonates with  $\text{ArTeNa}$ .** To a solution of diaryl ditelluride (0.5 mmol) in (1/1) (V/V) THF/EtOH (5 mL) was added sodium borohydride (1.2 mmol) in EtOH dropwise until the characteristic dark-red color of the ditelluride faded. Then 1-alkynylphosphonate (1 mmol) in THF (2 mL) was added, and the resulting solution was stirred overnight at room temperature, quenched with saturated aqueous  $\text{NH}_4\text{Cl}$ , and extracted with  $\text{CH}_2\text{Cl}_2$ . The organic phase was dried over sodium sulfate and concentrated under vacuum. The residue was purified by preparative TLC on silica gel (eluent:  $\text{AcOEt}$ :hexanes = 1:5) to give compounds **6a–f**.

(16) Gerard, J.; Bietlot, E.; Hevesi, L. *Tetrahedron Lett.* **1998**, *39*, 8735.

(17) Vinyl sulfones For reviews, See: (a) *The Chemistry of Sulphones and Sulphoxides*; Patai, S.; Rappoport, Z., Stirling, C. J. M., Eds.; Wiley: Chichester, 1988. (b) Lucchi, O. D.; Pasquato, L. *Tetrahedron* **1988**, *44*, 6755. (c) Block, E.; Aslam, M. *Tetrahedron* **1988**, *44*, 281. (e) Fuchs, P. L.; Braish, T. F. *Chem. Rev.* **1986**, *86*, 903.

(18) For a review on vinyl phosphonate, see: Minami, T.; Motoyoshiya, J. *Synthesis* **1992**, 333.

(19) Dang, H. P.; Linstrumelle, G. *Tetrahedron Lett.* **1978**, 191. (b) King, A. O.; Okukado, N.; Negishi, E. *J. Chem. Soc., Chem. Commun.* **1977**, 683.

(20) (a) Truce, W. E.; Wolf, G. C. *J. Org. Chem.* **1971**, *36*, 1727. (b) Chen, Z. M.; Trudell, M. L. *Synth. Commun.* **1994**, *24*, 3149.

(21) (a) Chattha, M. S.; Aguiar, A. M. *J. Org. Chem.* **1971**, *36*, 2719. (b) Hong, J. E.; Lee, C.-W.; Kwon, Y.; Oh, D. Y. *Synth. Commun.* **1996**, *26*, 1563.

(22) Iwata, N.; Morioka, T.; Kobayashi, T.; Asada, T.; Kinoshita, H.; Inomata, K. *Bull. Chem. Soc. Jpn.* **1992**, *65*, 1379.

(14) Kopchik, R. M.; Kampeier, J. A. *J. Am. Chem. Soc.* **1968**, *90*, 6733.

(15) Sung, J. W.; Park, C. P.; Gil, J. M.; Oh, D. Y. *Synth. Commun.* **1998**, *28*, 2635.



**(Z)-1-(Diethoxyphosphonyl)-2-(4-methylphenyltelluryl)-2-phenylethene (6a):** yield 0.27 g (59%); mp 95–96 °C;  $\nu$  (KBr)/cm<sup>-1</sup>: 3060, 2985, 2810, 1650, 1560, 1490, 1450, 1400, 1235, 840, 700, 610 cm<sup>-1</sup>;  $\delta_{\text{H}}$  (400 MHz) 7.28–7.30 (d,  $J = 7.9$ , 2H), 6.98 (m, 5H), 6.40–6.72 (d,  $J = 17.3$ , 1H), 4.16–4.23 (m, 4H), 2.19 (s, 3H), 1.37–1.41 (m, 6H); <sup>13</sup>C NMR  $\delta$  153.72 (d, <sup>2</sup> $J_{\text{P-C}} = 7.7$  Hz), 142.94 (d, <sup>3</sup> $J_{\text{P-C}} = 24.6$  Hz), 140.58, 137.83, 129.43, 127.74 (d, <sup>4</sup> $J_{\text{P-C}} = 1$  Hz), 127.53, 127.37, 119.97 (d, <sup>1</sup> $J_{\text{P-C}} = 189.9$  Hz), 114.01, 62.21 (d, <sup>2</sup> $J_{\text{P-C}} = 5$  Hz), 21.26, 16.56 (d, <sup>3</sup> $J_{\text{P-C}} = 6.1$  Hz); MS (EI)  $m/z$  460 (M<sup>+</sup>, 38.22), 369 (5.45), 239 (42.74), 195 (68.29), 183 (36.13), 167 (100), 109 (46.26), 91 (49.99); (Found: C, 49.66; H, 4.98. Calcd for C<sub>19</sub>H<sub>23</sub>O<sub>3</sub>PTe: C, 49.83; H 5.06%).

**(Z)-1-(Diethoxyphosphonyl)-2-(4-chlorophenyltelluryl)-2-phenylethene (6b):** yield 0.30 g (63%); mp 54–55 °C;  $\nu$  (KBr)/cm<sup>-1</sup> 3070, 2987, 1570, 1500, 1450, 1400, 1225, 1013, 835, 690, 610 cm<sup>-1</sup>;  $\delta_{\text{H}}$  (400 MHz) 6.83–7.43 (m, 9H), 6.23–6.53 (d,  $J = 17.1$ , 1H), 4.02–4.41 (m, 4H), 1.21–1.43 (m, 6H); <sup>13</sup>C NMR  $\delta$  153.46 (d, <sup>2</sup> $J_{\text{P-C}} = 7.6$  Hz), 142.54 (d, <sup>3</sup> $J_{\text{P-C}} = 24.5$  Hz), 141.83, 128.72, 127.84, 127.69 (d, <sup>4</sup> $J_{\text{P-C}} = 1.3$  Hz), 127.54, 120.02 (d, <sup>1</sup> $J_{\text{P-C}} = 189.2$  Hz), 115.99, 62.31 (d, <sup>2</sup> $J_{\text{P-C}} = 5.2$  Hz), 16.54 (d, <sup>3</sup> $J_{\text{P-C}} = 6.2$  Hz); MS (EI)  $m/z$  480 (M<sup>+</sup>, 20.83), 367 (6.48), 239 (47.61), 195 (78.72), 167 (100), 109 (47.61), 81 (33.63); (Found: C, 45.07; H, 4.13. Calcd for C<sub>18</sub>H<sub>20</sub>ClO<sub>3</sub>PTe: C, 45.19; H 4.21%).

**(Z)-1-(Diethoxyphosphonyl)-2-(phenyltelluryl)-2-phenylethene (6c):** yield 0.27 g (61%); mp 29–30 °C;  $\nu$  (KBr)/cm<sup>-1</sup>: 3100, 2985, 1560, 1480, 1450, 1400, 1230, 1020, 830, 690, 610 cm<sup>-1</sup>;  $\delta_{\text{H}}$  (400 MHz) 7.40–7.42 (m, 2H), 6.89–7.08 (m, 8H), 6.38–6.42 (d,  $J = 17.1$ , 1H), 4.16–4.23 (m, 4H), 1.38–1.41 (m, 6H); <sup>13</sup>C NMR  $\delta$  153.56 (d, <sup>2</sup> $J_{\text{P-C}} = 7.7$  Hz), 142.63 (d, <sup>3</sup> $J_{\text{P-C}} = 24.5$  Hz), 140.42, 128.38, 127.68, 127.62 (d, <sup>4</sup> $J_{\text{P-C}} = 1.1$  Hz), 127.50, 127.28, 119.83 (d, <sup>1</sup> $J_{\text{P-C}} = 189.7$  Hz), 117.91, 62.13 (d, <sup>2</sup> $J_{\text{P-C}} = 5.1$  Hz), 16.46 (d, <sup>3</sup> $J_{\text{P-C}} = 6.3$  Hz); MS (EI)  $m/z$  446 (M<sup>+</sup>, 40.94), 369 (35.24), 239 (44.68), 195 (69.72), 167 (100), 109 (43.34), 77 (32.94); (Found: C, 48.77; H, 4.73. Calcd for C<sub>18</sub>H<sub>21</sub>O<sub>3</sub>PTe: C, 48.70; H 4.77%).

**(Z)-1-(Diethoxyphosphonyl)-2-(4-fluorophenyltelluryl)-2-phenylethene (6d):** yield 0.31 g (67%); mp 39–40 °C;  $\nu$  (KBr)/cm<sup>-1</sup>: 3080, 2987, 1580, 1480, 1230, 1030, 830, 695, 600;  $\delta_{\text{H}}$  (400 MHz) 7.35–7.38 (m, 2H), 6.92–7.0 (m, 5H), 6.58–6.62 (m, 2H), 6.37–6.41 (d,  $J = 17.2$ , 1H), 4.17–4.21 (m, 4H), 1.38–1.41 (m, 6H); <sup>13</sup>C NMR  $\delta$  162.74 (d, <sup>1</sup> $J_{\text{F-C}} = 247$  Hz), 153.69 (d, <sup>2</sup> $J_{\text{P-C}} = 7.5$  Hz), 142.73 (d, <sup>3</sup> $J_{\text{F-C}} = 7.6$  Hz), 142.52 (d, <sup>3</sup> $J_{\text{P-C}} = 24.6$  Hz), 127.63, 127.57 (d, <sup>4</sup> $J_{\text{P-C}} = 1.3$  Hz), 127.41, 119.77 (d, <sup>1</sup> $J_{\text{P-C}} = 189.4$  Hz), 115.81 (d, <sup>2</sup> $J_{\text{F-C}} = 20.6$  Hz), 112.49 (d, <sup>4</sup> $J_{\text{F-C}} = 3.8$  Hz), 62.23 (d, <sup>2</sup> $J_{\text{P-C}} = 5.3$  Hz), 16.49 (d, <sup>3</sup> $J_{\text{P-C}} = 6$  Hz); MS (EI)  $m/z$  464 (M<sup>+</sup>, 37.78), 369 (10.57), 239 (47.43), 195 (75.99), 183 (36.03), 167 (100), 109 (46.73), 91 (49.99); (Found: C, 46.56; H, 4.43. Calcd for C<sub>18</sub>H<sub>20</sub>FO<sub>3</sub>PTe: C, 46.80; H 4.36%).

**(Z)-1-(Diethoxyphosphonato)-2-(4-methylphenyltelluryl)-1-heptene (6e):** yield 0.26 g (57%); oil;  $\nu$  (film)/cm<sup>-1</sup> 3030, 2950, 1570, 1500, 1465, 1400, 1235, 1040, 960, 800;  $\delta_{\text{H}}$  (400 MHz) 7.78–7.80 (m, 2H), 7.05–7.07 (m, 2H), 6.21–6.26 (d,  $J = 16.74$ , 1H), 4.08–4.13 (m, 4H), 2.36 (s, 3H), 2.17–2.20 (t,  $J = 7.84$ , 2H), 1.31–1.36 (m, 8H), 1.08–1.10 (m, 2H), 0.96–0.98 (m, 2H), 0.74–0.78 (t,  $J = 7.19$ , 3H); MS (EI)  $m/z$  454 (M<sup>+</sup>, 28.33), 363 (9.65), 233 (100), 221 (8.29), 205 (33.50), 177 (50.13), 95 (87.94); (Found: C, 47.57; H, 6.43. Calcd for C<sub>18</sub>H<sub>29</sub>PO<sub>3</sub>Te: C, 47.83; H 6.47%).

**(Z)-1-(Diethoxyphosphonyl)-2-(4-fluorophenyltelluryl)-1-heptene (6f):** yield 0.26 g (59%); Oil;  $\nu$  (film)/cm<sup>-1</sup> 3030, 2950, 1570, 1500, 1400, 1230, 1050, 960, 825;  $\delta_{\text{H}}$  (400 MHz) 7.87–7.91 (m, 2H), 6.93–6.98 (m, 3H), 6.24–6.28 (d,  $J = 16.51$ , 1H), 4.08–4.14 (m, 4H), 2.16–2.19 (t,  $J = 7.08$ , 2H), 0.131–1.37 (m, 8H), 1.09–1.12 (m, 2H), 0.98–1.00 (m, 2H), 0.76–0.80 (t,  $J = 7.19$ , 3H); <sup>13</sup>C NMR  $\delta$  163.47 (d, <sup>1</sup> $J_{\text{F-C}} = 248$  Hz), 155.31 (d, <sup>2</sup> $J_{\text{P-C}} = 6.1$  Hz), 144.18 (d, <sup>3</sup> $J_{\text{F-C}} = 7.8$  Hz), 116.60 (d, <sup>2</sup> $J_{\text{F-C}} = 20.5$  Hz), 115.85 (d, <sup>1</sup> $J_{\text{P-C}} = 192.4$  Hz), 110.34 (d, <sup>4</sup> $J_{\text{F-C}} = 3.7$  Hz), 61.94 (d, <sup>2</sup> $J_{\text{P-C}} = 5$  Hz), 42.43 (d, <sup>1</sup> $J_{\text{P-C}} = 23$  Hz), 30.83, 29.62 (d, <sup>4</sup> $J_{\text{P-C}} = 1.2$  Hz), 22.25, 16.45 (d, <sup>3</sup> $J_{\text{P-C}} = 6.3$  Hz), 13.90; MS (EI)  $m/z$  458 (M<sup>+</sup>, 27.82), 363 (17.09), 233 (75.13), 177 (44.59), 95 (100); (Found: C, 44.60; H, 5.71. Calcd for C<sub>17</sub>H<sub>26</sub>FO<sub>3</sub>PTe: C, 44.78; H 5.75%).

**Anti-Hydrotelluration of 1-Alkynyl Sulfones with ArTeNa.** To a solution of diaryl ditelluride (0.25 mmol) dissolved in (1/1) (V/V) THF/ethanol (2.5 mL) under N<sub>2</sub> was added a solution of sodium borohydride (0.6 mmol) in EtOH dropwise until the characteristic dark-red color of the ditelluride faded, and then 1-alkynyl sulfone (0.5 mmol) in THF was added. The resulting solution was stirred at room temperature for 20 min, quenched with saturated aqueous NH<sub>4</sub>Cl, and extracted with AcOEt. The organic phase was dried over sodium sulfate. Crude products (entries **7** and **8**) were concentrated under vacuum, and the residues were purified by preparative TLC on silica gel (AcOEt/hexanes = 1/10) to give compounds **8a,b**. Crude products (entries **9** and **10**) were treated with 1 equiv of Br<sub>2</sub> in AcOEt and concentrated under vacuum. The residues were purified by recrystallization with CH<sub>3</sub>OH/HCCl<sub>3</sub> to give products (**Z**)-**9a** and (**Z**)-**9b**.

**(Z)-1-(4-Methylphenylsulfonyl)-2-(phenyltelluryl)-2-phenylethene (8a):** yield 0.172 g (74%); mp 142–143 °C;  $\nu$  (KBr)/cm<sup>-1</sup> 2940, 1601, 1529, 1434, 1320, 1083, 810, 655;  $\delta_{\text{H}}$  (500 MHz): 7.95–7.97 (d,  $J = 8.2$ , 2H), 7.37–7.39 (m, 4H), 7.08–7.11 (m, 1H), 6.85–6.95 (m, 8H), 2.47 (s, 3H); <sup>13</sup>C NMR  $\delta$  147.94, 144.74, 140.80, 139.75, 138.09, 130.10, 130.03, 128.64, 128.26, 128.03, 127.87, 127.49, 127.43, 117.19, 21.78; MS (EI)  $m/z$  464 (M<sup>+</sup>, 16.96), 207 (16.15), 193 (43.96), 155 (45.62), 139 (26.50), 91 (100), 77 (53.45), 65 (19.11); (Found: C, 54.29; H, 3.86. Calcd for C<sub>21</sub>H<sub>18</sub>O<sub>2</sub>STe: C, 54.59; H, 3.93%).

**(Z)-1-(4-Methylphenylsulfonyl)-2-(4-fluorophenyltelluryl)-2-phenylethene (8b):** yield 0.180 g (75%); mp 144–145 °C;  $\nu$  (KBr)/cm<sup>-1</sup>: 2940, 1576, 1486, 1394, 11314, 1234, 1142, 1018, 810, 655;  $\delta_{\text{H}}$  (500 MHz): 7.94–7.95 (d,  $J = 8.2$ , 2H), 7.32–7.39 (m, 4H), 6.95–6.98 (m, 3H), 6.91 (s, 1H), 6.81–6.83 (m, 2H), 6.58–6.62 (t,  $J = 8.75$ , 2H), 2.47 (s, 3H); MS (EI)  $m/z$  482 (M<sup>+</sup>, 14.55), 225 (11.45), 193 (45.25), 155 (46.98), 139 (22.38), 91 (100); (Found: C, 52.34; H, 3.50. Calcd for C<sub>21</sub>H<sub>17</sub>FO<sub>2</sub>STe: C, 52.55; H 3.57%).

**(Z)-2-(4-Methylphenylsulfonyl)vinylphenyltellurium dibromides (9a):** yield 0.224 g (82%); mp 161–162 °C;  $\nu$  (KBr)/cm<sup>-1</sup>: 3002, 1594, 1436, 1295, 1257, 1140, 1078, 710;  $\delta_{\text{H}}$  (400 MHz): 8.26–8.28 (m, 3H), 8.06–8.08 (d,  $J = 8.32$ , 2H), 7.54–7.56 (m, 3H), 7.43–7.45 (d,  $J = 8.20$ , 2H), 6.95–6.97 (d,  $J = 8.69$ , 1H), 2.49 (s, 3H);  $\delta$  <sup>13</sup>C NMR 146.69, 134.64, 134.49, 134.14, 133.37, 131.81, 131.53, 130.48, 130.12, 129.13; MS (EI)  $m/z$  467 (M<sup>+</sup> – Br, 2.68), 388 (M<sup>+</sup> – 2Br, 56.85), 298 (17.69), 232 (24.60), 207 (43.64), 168 (29.80), 139 (41.76), 102 (35.29), 91 (65.35), 77 (100); (Found: C, 32.73; H, 2.52. Calcd for C<sub>15</sub>H<sub>14</sub>Br<sub>2</sub>O<sub>2</sub>STe: C, 33.01; H 2.59%).

**(Z)-2-(4-Methylphenylsulfonyl)-1-hexylvinyl phenyl tellurium dibromide (9b):** yield 0.224 g (71%); mp 127–128 °C;  $\delta_{\text{H}}$  (400 MHz): 8.64–8.66 (d,  $J = 7.23$ , 2H), 8.07–8.10 (d,  $J = 8.29$ , 2H), 7.55–7.61 (m, 3H), 7.41–7.43 (d,  $J = 8.21$ , 2H), 6.44 (s, 1H), 2.54–2.58 (m, 2H), 2.48 (s, 3H), 1.40–1.43 (m, 2H), 1.07–1.16 (m, 6H), 0.75–0.79 (t,  $J = 6.88$ , 3H);  $\delta$  <sup>13</sup>C NMR 153.31, 146.05, 137.45, 135.40, 131.99, 130.30, 130.10, 128.29, 126.96, 124.35, 35.91, 31.25, 28.54, 28.15, 22.32, 21.87, 13.95; MS (EI)  $m/z$  551 (M<sup>+</sup> – Br, 2.02), 472 (M<sup>+</sup> – 2Br, 31.47), 207 (36.55), 109 (100), 91 (71.51), 77 (56.99); (Found: C, 39.80; H, 4.06. Calcd for C<sub>21</sub>H<sub>26</sub>Br<sub>2</sub>O<sub>2</sub>STe: C, 40.04; H, 4.16%).

**Reaction of (E)-2-Iodo-1-alkenyl Sulfones with ArTeNa.** To a solution of diaryl ditelluride (0.25 mmol) dissolved in (1/1) (v/v) THF/ethanol (2.5 mL) under N<sub>2</sub> was added sodium borohydride (0.6 mmol) in EtOH dropwise until the characteristic dark-red color of the ditelluride faded, and then (E)-2-iodo-1-alkenyl sulfone (0.5 mmol) in THF was added at 0 °C. The resulting solution was stirred at room temperature for 20 min, quenched with saturated aqueous NH<sub>4</sub>Cl, and extracted with AcOEt. The organic phase was dried over sodium sulfate. Crude products **12a–f** (entries **11–16**) were treated with 1 equiv of Br<sub>2</sub> in AcOEt and concentrated under vacuum. The residues were purified via recrystallization with CH<sub>3</sub>OH/HCCl<sub>3</sub> to give (E)-**13a–f**. The crude product (entry **17**) was concentrated under vacuum, and the residue was purified by preparative TLC on silica gel (AcOEt:*n*-hexane = 1:10) to give compound (E)-**12g**.

**(E)-2-(4-Methylphenylsulfonyl)-1-butylvinyl phenyl tellurium dibromide (13a):** yield 0.217 g (72%); mp 149–150

$^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 2926, 1595, 1438, 1319, 1142, 1082, 736;  $\delta_{\text{H}}$  (500 MHz): 8.16–8.18 (m, 2H), 7.71–7.73 (d,  $J = 8.26$ , 2H), 7.57–7.59 (m, 1H), 7.47–7.50 (m, 2H), 7.36–7.37 (d,  $J = 8.0$ , 2H), 6.77 (s, 1H), 3.56–3.59 (t,  $J = 7.87$ , 2H), 2.46 (s, 3H), 1.61–1.65 (m, 2H), 1.45–1.48 (m, 2H), 0.94–0.97 (t,  $J = 7.37$ , 3H);  $\delta$   $^{13}\text{C}$  NMR 157.85, 145.49, 139.91, 137.10, 135.72, 132.46, 130.58, 130.20, 128.79, 127.60, 32.06, 31.38, 22.53, 21.79, 13.78; MS (EI)  $m/z$  523 ( $\text{M}^+ - \text{Br}$ , 3.89), 444 ( $\text{M}^+ - 2\text{Br}$ , 49.31), 237 (28.62), 207 (54.21), 139 (41.63), 91 (84.77), 81 (100); (Found: C, 37.77; H, 3.60; Calcd for  $\text{C}_{19}\text{H}_{22}\text{SO}_2\text{Br}_2\text{Te}$ : C, 37.92; H 3.68%).

**(E)-2-(4-Methylphenylsulfonyl)-1-butylvinyl 4-fluorophenyl tellurium dibromide (13b)**: yield 0.22 g (71%); mp 170–171  $^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 2930, 1577, 1486, 1390, 1313, 1234, 1142, 1081, 811, 777;  $\delta_{\text{H}}$  (500 MHz): 8.19–8.21 (m, 2H), 7.71–7.73 (d,  $J = 11.35$ , 2H), 7.37–7.38 (d,  $J = 7.73$ , 2H), 7.16–7.20 (m, 2H), 6.72 (s, 1H), 3.56–3.59 (t,  $J = 7.52$ , 2H), 2.47 (s, 3H), 1.58–1.66 (m, 2H), 1.44–1.48 (m, 2H), 0.94–0.97 (t,  $J = 7.14$ , 3H);  $\delta$   $^{13}\text{C}$  NMR 164.88 (d,  $^1J_{\text{F-C}} = 255.3$  Hz), 158.07, 145.59, 139.81, 138.33 (d,  $^3J_{\text{F-C}} = 8.8$  Hz), 136.94, 130.22, 127.58, 122.81 (d,  $^4J_{\text{F-C}} = 3.1$  Hz), 117.98 (d,  $^2J_{\text{F-C}} = 22$  Hz), 32.02, 31.29, 22.51, 21.78, 13.75; MS (EI)  $m/z$  541 ( $\text{M}^+ - \text{Br}$ , 8.88), 462 ( $\text{M}^+ - 2\text{Br}$ , 48.20), 225 (39.20), 155 (49.04), 139 (45.74), 95 (43.66), 91 (100), 81 (70.99); (Found: C, 36.70; H, 3.35. Calcd for  $\text{C}_{19}\text{H}_{21}\text{Br}_2\text{FO}_2\text{STe}$ : C, 36.82; H, 3.41%).

**(E)-2-(4-Methylphenylsulfonyl)-1-pentylvinyl phenyl tellurium dibromide (13c)**: yield 0.193 g (64%); mp 97–98  $^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 2926, 1595, 1438, 1319, 1143, 1081, 811, 736;  $\delta_{\text{H}}$  (500 MHz): 8.17–8.19 (m, 2H), 7.71–7.73 (d,  $J = 8.27$ , 2H), 7.58–7.59 (m, 1H), 7.47–7.50 (m, 2H), 7.35–7.37 (d,  $J = 8.17$ , 2H), 6.78 (s, 1H), 3.55–3.58 (t,  $J = 7.74$ , 2H), 2.46 (s, 3H), 1.63–1.66 (m, 2H), 1.34–1.41 (m, 4H), 0.88–0.92 (t,  $J = 6.94$ , 3H);  $\delta$   $^{13}\text{C}$  NMR 157.86, 145.41, 139.93, 137.04, 135.67, 132.37, 130.49, 130.12, 128.79, 127.51, 31.51, 31.28, 29.64, 22.22, 21.73, 13.86; MS (EI)  $m/z$  523 ( $\text{M}^+ - \text{Br}$ , 3.89), 444 ( $\text{M}^+ - 2\text{Br}$ , 49.31), 237 (28.62), 207 (54.21), 139 (41.63), 91 (84.77), 81 (100); (Found: C, 38.81; H, 3.88. Calcd for  $\text{C}_{20}\text{H}_{24}\text{Br}_2\text{O}_2\text{STe}$ : C, 39.00; H, 3.93%).

**(E)-2-(4-Methylphenylsulfonyl)-1-pentylvinyl 4-fluorophenyl tellurium dibromide (13d)**: yield 0.199 g (63%); mp 158–159  $^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 2930, 1578, 1486, 1393, 1313, 1234, 1141, 1081, 811, 653;  $\delta_{\text{H}}$  (500 MHz): 8.19–8.22 (m, 2H), 7.72–7.73 (d,  $J = 8.02$ , 2H), 7.57–7.59 (d,  $J = 8.02$ , 1H), 7.37–7.38 (m, 2H), 7.17–7.20 (m, 2H), 6.73 (s, 1H), 3.55–3.58 (t,  $J = 7.67$ , 2H), 2.47 (s, 3H), 1.63–1.66 (m, 2H), 1.34–1.40 (m, 4H), 0.89–0.92 (t,  $J = 6.62$ , 3H);  $\delta$   $^{13}\text{C}$  NMR 164.92 (d,  $^1J_{\text{F-C}} = 255.1$  Hz), 158.13, 145.57, 139.93, 138.32 (d,  $^3J_{\text{F-C}} = 8.8$  Hz), 137.05, 130.21, 127.61, 122.84, 117.98 (d,  $^2J_{\text{F-C}} = 22.1$ ), 31.49, 31.33, 29.69, 22.26, 21.75, 13.85; MS (EI)  $m/z$  553 ( $\text{M}^+ - \text{Br}$ , 1.14), 476 ( $\text{M}^+ - 2\text{Br}$ , 17.72), 225 (27.04), 157 (31.10), 95 (100), 91 (59.61); (Found: C, 37.77; H, 3.60. Calcd for  $\text{C}_{20}\text{H}_{23}\text{Br}_2\text{FO}_2\text{STe}$ : C, 37.90; H, 3.66%).

**(E)-2-(4-Methylphenylsulfonyl)-1-hexylvinyl 4-fluorophenyl tellurium dibromide (13e)**: yield 0.185 g (57%); mp 118–120  $^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 2926, 2854, 1578, 1486, 1394, 1314, 1234, 1141, 1081, 811, 654;  $\delta_{\text{H}}$  (500 MHz): 8.19–8.22 (m, 2H), 7.71–7.73 (m, 2H), 7.36–7.38 (m, 2H), 7.17–7.20 (m, 2H), 6.73 (s, 1H), 3.55–3.58 (m, 2H), 2.47 (s, 3H), 1.61–1.65 (m, 2H), 1.38–1.40 (m, 2H), 1.28–1.30 (m, 4H), 0.89–0.91 (m, 3H);  $\delta$   $^{13}\text{C}$  NMR 164.89 (d,  $^1J_{\text{F-C}} = 255.3$  Hz), 158.21, 145.57, 139.93, 138.34 (d,  $^3J_{\text{F-C}} = 8.7$  Hz), 137.02, 130.21, 127.60, 122.85 (d,  $^4J_{\text{F-C}} = 3.2$  Hz), 117.98 (d,  $^2J_{\text{F-C}} = 22.1$  Hz), 31.55, 31.34, 29.99, 28.93, 22.46, 21.77, 14.05; MS (EI)  $m/z$  569 ( $\text{M}^+ - \text{Br}$ , 2.60), 490 ( $\text{M}^+ - 2\text{Br}$ , 41.09), 265 (16.94), 225 (40.52), 155 (45.73), 139 (38.03), 109 (100), 91 (74.35); (Found: C, 38.73; H, 3.81. Calcd for  $\text{C}_{21}\text{H}_{25}\text{Br}_2\text{FO}_2\text{STe}$ : C, 38.93; H, 3.89%).

**(E)-2-(4-Methylphenylsulfonyl)-1-hexylvinyl phenyl tellurium dibromide (13f)**: yield 0.195 g (62%); mp 96–97  $^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 2924, 1595, 1437, 1319, 1143, 1081, 811, 736, 644, 539;  $\delta_{\text{H}}$  (500 MHz): 8.17–8.19 (m, 2H), 7.72–7.73 (m, 2H), 7.57–7.59 (m, 1H), 7.48–7.50 (m, 2H), 7.35–7.37 (m, 2H), 6.78 (s, 1H), 3.55–3.58 (m, 2H), 2.46 (s, 3H), 1.62–1.65 (m, 2H), 1.38–1.40 (m, 2H), 1.28–1.31 (m, 4H), 0.89–0.91 (m, 3H);  $\delta$   $^{13}\text{C}$  NMR 157.91, 145.40, 139.95, 137.08, 135.67, 132.38,

130.50, 130.12, 128.78, 127.52, 31.57, 31.30, 29.95, 28.88, 22.41, 21.72, 14.02; MS (EI)  $m/z$  551 ( $\text{M}^+ - \text{Br}$ , 2.22), 472 ( $\text{M}^+ - 2\text{Br}$ , 33.05), 265 (16.65), 207 (36.55), 157 (36.77), 109 (100), 91 (71.51), 77 (56.99); (Found: C, 39.88; H, 4.11. Calcd for  $\text{C}_{21}\text{H}_{26}\text{Br}_2\text{O}_2\text{STe}$ : C, 40.04; H, 4.16%).

**(E)-1-(4-Methylphenylsulfonyl)-2-phenyltelluryl-2-phenylethene (12g)**: yield 0.146 g (63%); mp 129–130  $^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 3046, 1576, 1435, 1322, 1140, 1083, 769;  $\delta_{\text{H}}$  (500 MHz): 7.77–7.79 (m, 2H), 7.42 (m, 1H), 7.30–7.32 (m, 4H), 7.26–7.20 (m, 3H), 7.09–7.12 (m, 4H), 6.39 (s, 1H), 2.36 (s, 1H);  $\delta$   $^{13}\text{C}$  NMR 144.90, 143.66, 140.83, 138.68, 137.26, 132.30, 130.29, 129.92, 129.34, 128.66, 127.83, 127.57, 127.41, 114.84, 21.59; MS (EI)  $m/z$  462 ( $\text{M}^+$ , 16.36), 309 (7.44), 207 (24.14), 155 (52.24), 105 (24.68), 91 (100), 77 (66.71); (Found: C, 54.12; H, 3.87. Calcd for  $\text{C}_{21}\text{H}_{18}\text{O}_2\text{STe}$ : C, 54.59; H, 3.93%).

**Anti-Tellurosulfonates of 1-Alkynes.** A solution of alkyne (1.1 mmol), sodium arylsulfinate (5 mmol), and diorgano ditelluride (0.5 mmol) in 5 mL of 80% aqueous acetic acid was heated at 80  $^{\circ}\text{C}$  for the time indicated in Table 3. The reaction mixture was diluted with 20 mL of AcOEt, washed with saturated aqueous sodium bicarbonate, dried over  $\text{Na}_2\text{SO}_4$ , and concentrated in a vacuum. Crude products (entires **22**, **23**, and **26**) were concentrated under vacuum, and the residues were purified by preparative TLC on silica gel (AcOEt:hexanes = 1:10) to give compounds **12g**, **12h**, and **12k**. Crude products (entries **18–21**, and **24** and **25**) were treated with 1 equiv of  $\text{Br}_2$  in AcOEt and concentrated under vacuum. The residues were purified by recrystallization with  $\text{CH}_3\text{OH}/\text{HCl}_3$ .

**(E)-1-(4-Chlorophenylsulfonyl)-2-phenyltelluryl-2-phenylethene (12h)**: yield 0.412 g (85%); mp 154–155  $^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 3020, 1576, 1326, 1142, 1083, 776, 605, 540;  $\delta_{\text{H}}$  (500 MHz): 7.80–7.82 (m, 2H), 7.43–7.45 (m, 1H), 7.31–7.35 (m, 4H), 7.20–7.24 (m, 3H), 7.08–7.10 (m, 2H), 6.38 (s, 1H);  $\delta$   $^{13}\text{C}$  NMR 146.57, 140.94, 140.17, 139.46, 137.13, 131.91, 130.45, 130.13, 129.09, 128.98, 128.93, 128.02, 127.45, 114.81; MS (EI)  $m/z$  484 ( $\text{M}^+$ , 25.01), 309 (11.57), 277 (7.24), 213 (43.59), 205 (32.02), 175 (36.40), 111 (59.97), 77 (100); (Found: C, 49.69; H, 3.08. Calcd for  $\text{C}_{20}\text{H}_{15}\text{ClO}_2\text{STe}$ : C, 49.79; H, 3.13%).

**(E)-2-(Phenylsulfonyl)-1-butylvinyl phenyl tellurium dibromide (13i)**: yield 0.300 g (51%); mp 135–136  $^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 3057, 2930, 1436, 1390, 1147, 1081, 735, 659;  $\delta_{\text{H}}$  (500 MHz): 8.19–8.21 (m, 2H), 7.87–7.88 (m, 2H), 7.69–7.72 (m, 1H), 7.59–7.62 (m, 3H), 7.49–7.52 (m, 2H), 6.79 (s, 1H), 3.60–3.63 (m, 2H), 1.64–1.68 (m, 2H), 1.46–1.50 (m, 2H), 0.96–0.99 (m, 3H);  $\delta$   $^{13}\text{C}$  NMR 158.33, 139.93, 139.51, 135.66, 134.22, 132.43, 130.52, 129.53, 128.87, 127.45, 31.99, 31.39, 22.46, 13.73; MS (EI)  $m/z$  509 ( $\text{M}^+ - \text{Br}$ , 2.98), 430 ( $\text{M}^+ - 2\text{Br}$ , 28.58), 223 (19.10), 207 (38.51), 143 (30.67), 77 (100); (Found: C, 36.66; H, 3.32; Calcd for  $\text{C}_{18}\text{H}_{20}\text{Br}_2\text{O}_2\text{STe}$ : C, 36.78; H, 3.43%).

**(E)-2-(Phenylsulfonyl)-1-hexylvinyl phenyl tellurium dibromide (13j)**: yield 0.376 g (61%); mp 96–98  $^{\circ}\text{C}$ ;  $\nu$  (KBr)/ $\text{cm}^{-1}$ : 3032, 2925, 1446, 1326, 1147, 1081, 792, 635;  $\delta_{\text{H}}$  (500 MHz): 8.17–8.18 (m, 2H), 7.84–7.86 (m, 2H), 7.66–7.69 (m, 1H), 7.56–7.59 (m, 3H), 7.47–7.50 (m, 2H), 6.78 (s, 1H), 3.56–3.59 (m, 2H), 1.62–1.68 (m, 2H), 1.38–1.42 (m, 2H), 1.29–1.31 (m, 4H), 0.89–0.91 (m, 3H);  $\delta$   $^{13}\text{C}$  NMR 158.52, 140.17, 139.70, 135.71, 134.27, 132.52, 130.65, 129.60, 128.75, 127.58, 131.70, 31.39, 30.07, 28.99, 22.49, 14.07; MS (EI)  $m/z$  537 ( $\text{M}^+ - \text{Br}$ , 2.32), 458 ( $\text{M}^+ - 2\text{Br}$ , 35.20), 207 (39.46), 143 (40.32), 109 (97.88), 77 (100); (Found: C, 38.85; H, 3.84. Calcd for  $\text{C}_{20}\text{H}_{24}\text{Br}_2\text{O}_2\text{STe}$ : C, 39.00; H, 3.93%).

**(E)-1-(4-Methylphenylsulfonyl)-2-(phenyltelluryl)-ethene (12k)**: yield 0.1 g (26%); mp 124–126  $^{\circ}\text{C}$ ;  $\delta_{\text{H}}$  (400 MHz): 8.57–8.61 (d,  $J = 15.83$ , 1H), 7.76–7.78 (m, 2H), 7.69–7.72 (m, 2H), 7.41–7.43 (m, 1H), 7.30–7.33 (m, 4H), 6.41–6.45 (d,  $J = 15.82$ , 1H), 2.42 (s, 3H);  $\delta$   $^{13}\text{C}$  NMR 144.32, 140.09, 137.48, 134.77, 130.29, 130.01, 129.72, 127.76, 126.15, 110.83, 21.69; MS (EI)  $m/z$  388 ( $\text{M}^+$ , 50.83), 207 (40.66), 91 (65.35), 77 (100); (Found: C, 46.42; H, 3.56; Calcd for  $\text{C}_{15}\text{H}_{14}\text{O}_2\text{STe}$ : C, 46.68; H, 3.66%).

**Anti-Addition of ArTel with 1-Alkynes.** A mixture of 1 mmol of ArTel (prepared by the reaction of a diarylditelluride (0.5 mmol) with  $\text{I}_2$  (0.55 mmol) in THF at 0  $^{\circ}\text{C}$ ) and 1 mmol of an alkyne was stirred at room temperature for 12 h followed

by the treatment with saturated  $\text{Na}_2\text{S}_2\text{O}_3$ . After stirring for 30 min, the mixture was extracted with hexanes, dried over  $\text{CaCl}_2$ , and concentrated in a vacuum. The residues were purified by preparative TLC on silica gel eluted with hexanes to give compounds **16**, which reacted with  $\text{Br}_2$  (1 equiv) to afford (*E*)-**17a-f**.

**(E)-(2-Iodo-1-butyl)vinyl phenyl tellurium dibromide (17a)**: yield 0.258 g (45%); mp 146–147 °C;  $\nu$  (film)/ $\text{cm}^{-1}$ : 3067, 2928, 1636, 1438, 1253, 1112, 994, 730, 680;  $\delta_{\text{H}}$  (400 MHz) 8.29–8.32 (m, 2H), 7.54–7.61 (m, 3H), 7.37 (s, 1H), 3.10–3.14 (t,  $J$  7.92, 2H), 1.72–1.77 (m, 2H), 1.48–1.54 (m, 2H), 0.98–1.01 (t,  $J$  = 7.3, 3H);  $\delta$   $^{13}\text{C}$  NMR 144.52, 135.74, 132.19, 130.51, 130.39, 129.40, 95.52, 38.00, 30.72, 22.28, 13.90; MS (EI)  $m/z$ : 495 ( $\text{M}^+$  – 79, 4.32), 416 (17.03), 334 (10.95), 207 (53.19), 77 (100). (Found: C, 24.96; H, 2.55. Calcd for  $\text{C}_{12}\text{H}_{15}\text{Br}_2\text{ITe}$ : C, 25.13; H 2.64%).

**(E)-(2-Iodo-1-butyl)vinyl 4-chlorophenyl tellurium dibromide (17b)**: yield 0.322 g (53%); mp. 123–124 °C;  $\nu$  (film)/ $\text{cm}^{-1}$ : 3065, 2955, 1636, 1559, 1471, 1385, 1094, 1088, 1003, 807, 702;  $\delta_{\text{H}}$  (400 MHz) 8.25–8.27 (m, 2H), 7.52–7.54 (m, 2H), 7.35 (s, 1H), 3.09–3.13 (t,  $J$  = 7.98, 2H), 1.71–1.75 (m, 2H), 1.48–1.54 (m, 2H), 0.98–1.01 (t,  $J$  = 7.3, 3H);  $\delta$   $^{13}\text{C}$  NMR 144.58, 139.12, 137.06, 130.59, 127.04, 95.68, 37.87, 30.68, 22.23, 13.89; MS (EI)  $m/z$ : 529 ( $\text{M}^+$  – 79, 4.69), 450 (26.66), 368 (17.00), 241 (87.09), 81 (100). (Found: C, 23.83; H, 2.28. Calcd for  $\text{C}_{12}\text{H}_{14}\text{Br}_2\text{ClITe}$ : C, 23.71; H 2.32%).

**(E)-(2-Iodo-1-pentyl)vinyl phenyl tellurium dibromide (17c)**: yield 0.282 g (48%); mp 93–94 °C;  $\nu$  (film)/ $\text{cm}^{-1}$ : 3054, 2930, 2984, 1636, 1584, 1437, 1116, 994, 729, 679;  $\delta_{\text{H}}$  (400 MHz) 8.30–8.32 (m, 2H), 7.54–7.61 (m, 3H), 7.37 (s, 1H), 3.09–3.13 (t,  $J$  = 7.86, 2H), 1.75–1.78 (m, 2H), 1.39–1.49 (m, 4H), 0.92–0.96 (t,  $J$  = 7.0, 3H);  $\delta$   $^{13}\text{C}$  NMR 144.47, 135.73, 132.15, 130.46, 129.37, 95.63, 38.16, 31.10, 28.03, 22.40, 14.00; MS (EI)  $m/z$ : 509 ( $\text{M}^+$  – 79, 7.36), 430 (16.44), 334 (4.85), 207 (38.17), 95 (100). (Found: C, 26.88; H, 2.92. Calcd for  $\text{C}_{13}\text{H}_{17}\text{Br}_2\text{ITe}$ : C, 26.57; H 2.92%).

**(E)-(2-Iodo-1-pentyl)vinyl 4-chlorophenyl tellurium dibromide (17d)**: yield 0.317 g (51%); mp 104–105 °C;  $\nu$  (film)/ $\text{cm}^{-1}$ : 3054, 2931, 2869, 1560, 1472, 1385, 1179, 1093, 1004, 806, 721, 655;  $\delta_{\text{H}}$  (400 MHz) 8.25–8.27 (m, 2H), 7.52–

7.54 (m, 2H), 7.36 (s, 1H), 3.09–3.12 (t,  $J$  = 7.63, 2H), 1.73–1.77 (m, 2H), 1.39–1.48 (m, 4H), 0.92–0.96 (t,  $J$  = 7.16, 3H);  $\delta$   $^{13}\text{C}$  NMR 144.62, 139.16, 137.05, 130.62, 127.08, 95.66, 38.08, 31.10, 28.33, 22.41, 13.97; MS (EI)  $m/z$ : 543 ( $\text{M}^+$  – 79, 11.90), 464 (31.12), 241 (40.27), 95 (100); (Found: C, 25.17; H, 2.55. Calcd for  $\text{C}_{13}\text{H}_{16}\text{Br}_2\text{ClITe}$ : C, 25.10; H 2.59%).

**(E)-(2-Iodo-1-hexyl)vinyl phenyl tellurium dibromide (17e)**: yield 0.222 g (37%); mp 72–73 °C;  $\nu$  (film)/ $\text{cm}^{-1}$ : 3056, 2925, 2850, 1582, 1437, 994, 728, 678;  $\delta_{\text{H}}$  (400 MHz) 8.30–8.32 (m, 2H), 7.54–7.61 (m, 3H), 7.37 (s, 1H), 3.09–3.13 (t,  $J$  = 7.78, 2H), 1.74–1.78 (m, 2H), 1.46–1.50 (m, 2H), 1.34–1.36 (m, 4H), 0.90–0.93 (t,  $J$  = 6.72, 3H);  $\delta$   $^{13}\text{C}$  NMR 144.53, 135.73, 132.19, 130.51, 129.43, 95.53, 38.25, 31.51, 28.73, 28.64, 22.57, 14.12; MS (EI)  $m/z$ : 523 ( $\text{M}^+$  – 79, 2.36), 442 (12.09), 334 (7.27), 207 (40.43), 109 (100); (Found: C, 27.85; H, 3.14. Calcd for  $\text{C}_{14}\text{H}_{19}\text{Br}_2\text{ITe}$ : C, 27.95; H 3.18%).

**(E)-(2-Iodo-1-hexyl)vinyl 4-chlorophenyl tellurium dibromide (17f)**: yield 0.273 g (43%); mp. 80–81 °C;  $\nu$  (film)/ $\text{cm}^{-1}$ : 3057, 2925, 2950, 1571, 1560, 1472, 1385, 1091, 1002, 804, 720;  $\delta_{\text{H}}$  (400 MHz) 8.25–8.27 (m, 2H), 7.52–7.54 (m, 2H), 7.36 (s, 1H), 3.09–3.12 (t,  $J$  = 7.54, 2H), 1.72–1.76 (m, 2H), 1.46–1.49 (m, 2H), 1.33–1.37 (m, 4H), 0.90–0.93 (t,  $J$  = 6.89, 3H);  $\delta$   $^{13}\text{C}$  NMR 144.65, 139.21, 137.05, 130.65, 127.08, 95.58, 38.15, 31.49, 28.71, 28.64, 22.56, 14.12; MS (EI)  $m/z$ : 557 ( $\text{M}^+$  – 79, 4.83), 478 (22.96), 241 (55.30), 109 (100); (Found: C, 26.41; H, 2.81. Calcd for  $\text{C}_{14}\text{H}_{18}\text{Br}_2\text{ClITe}$ : C, 26.44; H 2.85%).

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**Supporting Information Available:**  $^1\text{H}$  NMR spectra of **6a-f**, **8a,b**, **9a,b**, **12g**, **12h**, **12k**, **13a-f**, **13i**, **13j**, and **17a-f**;  $^{13}\text{C}$  NMR spectra of **6a-d**, **6f**, **8a**, **9a,b**, **12g**, **12h**, **12k**, **13a-f**, **13i-j**, and **17a-f**;  $^1\text{H}$ – $^1\text{H}$  NOESY spectra of **6c**, **6e**, **8b**, **9b**, and **13f**, and ORTEP figures of **6a** and **17b**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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